

## **Applications of Powder Diffraction**

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# Advantages of synchrotron powder diffraction

#### **High intensity**

Small samples (powder or single crystal)

High speed (fast in situ processes, unstable compounds...)

High signal-to-noise  $\rightarrow$  better data

→ weak features (impurities, satellite reflections, diffuse scattering)

High resolution  $\rightarrow$  complex structures with big unit cells

→ fine peak splitting (phase transitions, decomposition)

 $\rightarrow$  peak shapes defined by sample properties (strain, particle size)

#### **Tunable energy**

Highlight elements of interest

Minimise absorption

#### **High energy**

High  $Q \rightarrow$  more accurate ADPs, potential for PDF analysis

Compressed patterns  $\rightarrow$  can use restrictive sample environments (pressure, fields)

High penetration  $\rightarrow$  minimise surface effects

#### **Detectors**



#### Multi Analyser Crystal Detectors

Extremely high resolution but slow

#### Strip detectors:

High resolution and fast but poor energy discrimination e.g. Mythen

#### Area detectors:

Lower resolution but provide texture information e.g. Mar CCD Mar 345 image plate Eiger











# **High Resolution**





2006: Jarosite minerals detected by the Mössbauer instrument on the *Opportunity* rover in Meridiani Planum.

#### Why is this significant?

• Jarosite is the first hydrated mineral to be positively identified on the surface of Mars.

Jarosites need a "wet" environment to form but an arid one to persist.

• Jarosite is crucial to understanding the role of water in the environmental evolution of Mars.







#### **Bioheap leaching**

Jarosite forms a kinetic barrier to the desired reaction.

#### Zinc Smelting

Jarosite formation is a way of removing Iron impurities from solution.











#### B. Monoclinic Jarosite







### This is what we initially expected...

# Rhombohedral Jarosite





#### This is what we got...



#### Too much resolution?



Caused by crystallisation on capillary walls

• Effectively two displaced samples

Artefact modelling in diffraction software necessary to account for heterogeneous nucleation on capillary walls

Broad applicability for nucleation studies



# **Energy Selection**





#### Yttrium containing compound at 18 keV







## Signal to noise

#### (We like tiny samples)





### Tiny samples





## Lab Sample

- Large amount of material required
- Sample loading can produce texture effects

## Synchrotron Sample



#### **Tiny samples**

#### LETTER

doi:10.1038/nature17678

## Ancient micrometeorites suggestive of an oxygen-rich Archaean upper atmosphere

Andrew G. Tomkins<sup>1</sup>, Lara Bowlt<sup>1</sup>, Matthew Genge<sup>2,3</sup>, Siobhan A. Wilson<sup>1</sup>, Helen E. A. Brand<sup>4</sup> & Jeremy L. Wykes<sup>1,4,5</sup>













#### **Tiny samples**



The first Australian Synchrotron powder diffraction analysis of pigment from a Wandjina motif in the Kimberley, Western Australia

Jillian Huntley<sup>1</sup>, Helen Brand<sup>2</sup>, Maxime Aubert<sup>3</sup> and Michael J. Morwood<sup>3</sup>







### Infinite building blocks





# Designing thermoelectric Materials.

Small changes in chemistry can have big changes in physical properties.

These small structural changes can only be mapped with the highest signal to noise.





# Throughput











## Speed









#### In situ studies: phase evolution







## Sodium Distribution and Reaction Mechanisms of a $Na_3V_2O_2(PO_4)_2F$ Electrode during Use in a Sodium-Ion Battery

N. Sharma, P. Serras, V. Palomares, H. E. A. Brand, J. Alonso, P. Kubiak, M.L. Fdez-Gubieda and T. Rojo

Published: Chem. Mater. 2014, 26, 3391-3402

- Na-ion batteries offer a more economical alternative to Liion batteries because Na is globally more abundant than Li, but...
- New electrode materials to improve the energy density and reversibility of Na insertion/extraction processes
- *In situ* charging and discharging using a modified CR2032 cell to observe changes in structure at the cathode.







- Key features include an initial region in the charging profile where at least three isostructural phases are formed with varying amounts of Na in each phase
- These transform via two two-phase reaction mechanisms both on charging and discharging.
- There are significant differences in behaviour during charging and discharging
- Work continues...





## **Complex sample environments**







#### Capillary sample environments

High Temperature (RT – 1500° C)

Low Temperature (80 - 480 K)

Gas environments Flowing or static – 20 bar\*





# Temperature controlled invertible selectivity for adsorption of $N_2$ and $CH_4$ by molecular trapdoor chabazites

J. Shang, G. Li, Q. Gu, R. Singh, P. Xiao, J. Z. Liu and P. A. Webley

Published: Chem. Commun., 2014, 50, 4544

- N<sub>2</sub> in CH<sub>4</sub> gas streams presents a problem for the methane economy
- A molecular trap door gas separation mechanism in the chabazite zeolite has been identified.
- Chabazite consists of an 8-member ring with additional 'door keeping' cations (K<sup>+</sup>) that block the path for gas molecules and prevent them entering the ring
- Thermal vibration can cause deviation of the 'door keeping' cation to allow the gas molecule to pass into the ring







#### Other sample environments



#### User defined sample environments





## **Complementary techniques**













synchrotron.org.au

Jupiter www.jpl.nasa.gov Image Credit: NASA/JPL/University of Arizona



#### Simultaneous synchrotron X-ray powder diffraction and Near IR characterisation





# When is a synchrotron not necessarily the answer?

# When can you get comparable results from a lab machine?







### **Poorly crystalline material**





#### **Clay Mineral Structure**



Clay is layered. The layers can be repeated or multiply stacked.

There can be cations and water in between the layers, or in between some layers and not others.

This causes the distances between similar planes of atoms to vary throughout the structure.





### **Nanomaterials**





## **Complex high-Z element mixes**







## Non – ideal geometry









Al2O3 in a capillary

Nice, sharp peaks, sensible intensities

Good powder averaging



Broadened peak shapes

Different intensities

Poor powder averaging



## Talk to your beamline scientist!

# (if you are the beamline scientist, know your instrument!)







## **Questions?**



